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# Influence of the Electron Selective Contact on the Interfacial Recombination in Fresh and Aged Perovskite Solar Cells

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**Abstract:** In this work, we have used TiO<sub>2</sub> and SnO<sub>2</sub> layers as electron selective contact (ESC) in ni-p perovskite solar cells configuration. To study and compare the ion migration kinetics of these ESC, CsFAMAPbIBr and MAPbI<sub>3</sub>-based devices were fabricated and characterised in fresh (1 day) and aged (28 days) conditions. Depending on the ESC and perovskite composition, devices reveal a different progression over time in terms of hysteresis and performance. Using transient photovoltage (TPV) and transient photocurrent (TPC) techniques, we studied the kinetics of carrier extraction and recombination, which showed that aged devices present slower recombination kinetics compared to their fresh counterparts, revealing a positive effect of the aging process. Finally, transient of the transient, derived from the TPV technique, discloses that TiO<sub>2</sub> accumulates more charges in the ESC/perovskite interface compared to SnO<sub>2</sub> and that the ion migration kinetics are directly related to the perovskite composition.

**Keywords:** electron selective contact; TiO<sub>2</sub>; SnO<sub>2</sub>; stability; perovskite; interfacial recombination; transient photovoltage; transient of the transient

## 1. Introduction

In a typical n-i-p device configuration in perovskite solar cells (PSCs), the right selection of an electron selective contact (ESC) that selectively extracts electrons and prevents hole injection is key to obtain efficient devices [1]. While there is intense research to find suitable n-type materials, TiO<sub>2</sub> and SnO<sub>2</sub> are still the most used materials [2,3]. Several studies have shown that the performances of PSCs with SnO<sub>2</sub> are better than with TiO<sub>2</sub>, especially in device stability. Additionally, SnO<sub>2</sub> shows better optical and electrical properties, favourable band alignment and low trap density [4],[5]. Thus, to increase solar cells performance, it is key to understand the major recombination losses [6]. Among others, interfacial recombination between perovskite and the selective contacts has been identified as one of the major carrier losses affecting device performance [6,7]. Moreover, it has been demonstrated that the underlying mechanisms governing the device performance can vary significantly depending on the perovskite composition in contact with the ESC [8],[9].

In previous papers, we further reported carrier losses mechanisms in different types of solar cells by using advanced transient techniques — charge extraction (CE), transient photovoltage (TPV) and transient photocurrent (TPC) [10–12]. Particularly, for perovskite devices, we have used a modification of TPV named "transient of the transient" (TROTR) [13,14]. In a typical TPV measurement, the device is illuminated until its open-circuit voltage (Voc) stabilises. Then, a laser pulse creates a small perturbation in the Voc. This laser pulse generates an extra number of charges that are forced to recombine as the device continues under open-circuit conditions [15]. In the modified TPV technique, a negative transient deflection appears at the early stages of illumination, when the Voc has not

reached the equilibrium. This negative signal can provide information about ionic motion and carrier recombination [16,17]. Based on our previous works [17,18], where we analysed the behaviour of the TROTR for fresh and aged PSCs, we concluded that the influence of ion migration on carrier lifetime is reduced over the different measurement days, mainly because of oxygen having an influence on both the perovskite and the hole selective contact material (HSC) [17]. In addition, we described different ionic migration behaviours depending on the perovskite composition, being more important in I-based perovskite compared to Br-based perovskites [18].

In this work, we focused on the influence of the electron selective contact  $-\text{TiO}_2$  and  $\text{SnO}_2$ —over the ion migration kinetics and carrier recombination in both fresh (1 day) and aged (28 days) devices using different perovskite compositions: methylammonium iodide (MAPbI<sub>3</sub>) and Cs<sub>0.1</sub>(FA<sub>0.85</sub>MA<sub>0.15</sub>)<sub>0.9</sub>Pb(I<sub>0.85</sub>Br<sub>0.15</sub>)<sub>3</sub>, herein referred to as CsFAMAPbIBr (FA = CH(NH<sub>2</sub>)<sub>2</sub><sup>+</sup>; MA = CH<sub>3</sub>NH<sub>3</sub><sup>+</sup>), also known as the triple cation perovskite. We studied the changes in device performance and carrier kinetics over the different measurement days, first, comparing both ESC for the triple cation perovskite, and second, using the SnO<sub>2</sub> as ESC for both perovskite compositions.

The results obtained revealed that both ESC and perovskite composition have a strong influence on the ionic dynamics that directly impact the stability and the performance in PSCs.

#### 2. Materials and Methods

*Device fabrication*: fluorine tin oxide (FTO)-coated glasses were cleaned with Hellmanex<sup>™</sup> soap solution in water using a toothbrush, then, they were ultrasonicated in distilled water, acetone and isopropanol. Finally, they were dried with an air flow and treated with UV/O<sub>3</sub> for 30 min.

Compact and mesoporous TiO<sub>2</sub> layers were deposited onto the clean FTO substrates by spin-coating 0.3 M Ti(iPrO)<sub>2</sub>(acac)<sub>2</sub> solution in 2-propanol using 4000 rpm, 1000 rpm/s for 25 s. The substrates were dried at 120 °C for 5 min, followed by thermal annealing at 450 °C for 30 min. Once the substrates were cold, they were plunged into a solution of 40 mM TiCl<sub>4</sub> at 70 °C for 30 min. Then, they were rinsed with distilled water and ethanol and dried with a strong air flow. The mesoporous TiO<sub>2</sub> layer was deposited by using a a commercial paste (30NR-D, Greatcell) with a 30 nm particle size diluted in ethanol (1:7 weight ratio) and spin-coated at 6000 rpm, 1000 rpm/s for 30 s. Then, they were heated by using the same conditions as compact TiO<sub>2</sub> layer.

A compact SnO<sub>2</sub> layer was grown by chemical bath deposition (CBD) technique on top of the FTO substrates. The solution was prepared dissolving first 0.75 g of urea in 60 mL of deionised H<sub>2</sub>O. Then, 15  $\mu$ L of mercaptopropionic acid and 750  $\mu$ L of HCl (37 wt%) were added. Finally, SnCl<sub>2</sub> · 2H<sub>2</sub>O was added to the solution to a final concentration of 0.012 M (162 mg). The solution was stirred for 2 min and transferred to a glass container were the substrates are immersed vertically. The deposition was made by putting them in a lab oven at 70 °C for 3 h. Then, the substrates were transferred to another glass container with H<sub>2</sub>O, and they were ultrasonicated for 2 min in order to remove any loosely bound material. Then, the substrates were dried and annealed at 180 °C for 1 h. Prior to perovskite deposition, SnO<sub>2</sub> films must be treated with UV/O<sub>3</sub> for 30 min.

Methylammonium lead iodide (MAPbI<sub>3</sub>) was deposited by performing two steps: 80  $\mu$ L of a 1 M PbI<sub>2</sub> solution in DMF:DMSO (9:1, vol%) was spin-coated at 2000 rpm, with 2000 rpm/s for 90 s, and after 60 s, 100  $\mu$ L of a 50 mg/mL MAI solution in IPA was spin-coated on top of the PbI<sub>2</sub> while spinning. The samples were treated by thermal annealing for 45 min at 100 °C.

Triple cation perovskite, CsFAMAPbIBr, was prepared by dissolving 1.2 M PbI<sub>2</sub>, 1.04 M formamidinium iodide (FAI), 0.16 M PbBr<sub>2</sub>, 0.16 M methylammonium bromide (MABr), and 0.1 M of CsI into DMF:DMSO (4:1, volume ratio). To help the dissolution, it was stirred at 60 °C for 30 min. Then, the solution was filtered and 40  $\mu$ L were spin-coated with a two-step program, first 2000 rpm, 500 rpm/s and 12 s, and then 5000 rpm 500 rpm/s 25 s. Ten seconds before the end of the spinning process, 100  $\mu$ L of chlorobenzene were

dropped in the centre of the substrate while spinning. Then, the films were annealed at 100  $^{\circ}C/1$  h. We prepared the devices with a continuous N<sub>2</sub> flow to remove the solvent atmosphere that can affect the perovskite morphology.

After that, a 60 mM spiro-OMeTAD solution in chlorobenzene with 4-tert-butylpiridine (TBP), LiTFSI (1.8 M solution in acetonitrile) and Co(III)TFSI (0.25 M solution in acetonitrile) as dopants was prepared as hole selective contact. The molar ratio employed was 1 spiro-OMeTAD:3.3 TBP:0.5 LiTFSI:0.05 Co (III) TFSI. Then, 35  $\mu$ L of this solution were spin-coated at 4000 rpm, 2000 rpm/s for 30 s on top of the perovskite layer. The devices were completed by evaporating 80 nm of Au under high-vacuum conditions.

*Aging process*: Devices were stored in the dark under dry conditions (<10% H<sub>2</sub>O, obtained via a synthetic airflow) for up to 6 weeks. The devices were fully characterised during this period to determine the effects of aging on the different device configurations.

*Characterization techniques:* The solar cells performance was obtained by using a Solar Simulator (ABET) and a source meter (Keithley 2400). The J-V curves of devices — with an active area of 0.09 cm<sup>2</sup> and scan rate of 0.04 V/s — were recorded at 1 Sun conditions (100 mW/cm<sup>2</sup>, AM 1.5G) previously calibrated with a Si cell as a reference. The charge extraction (CE) and transient photovoltage (TPV) analyses were carried out by a white LED controlled by a power supply and a control box that swap from open to short-circuit states. All the decays were registered in an oscilloscope Yokogawa DLM2052 by recording drops in voltage. Light perturbations pulses for TPV were generated by a nanosecond PTI GL-3300 nitrogen laser at 590 nm.

# 3. Results and Discussion

#### 3.1. SnO<sub>2</sub> vs. TiO<sub>2</sub> for CsFAMAPIBr Based Devices

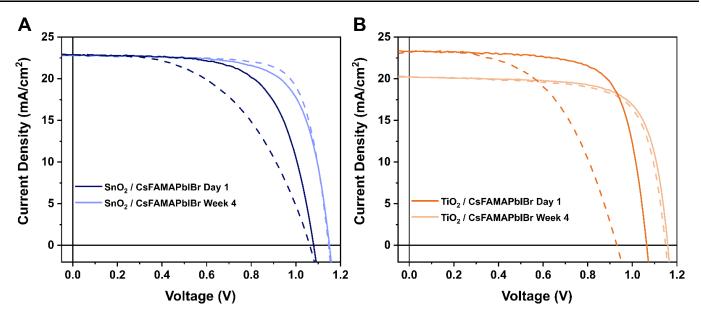
First, we studied the photovoltaic parameters of the solar cells with the different ESC (SnO<sub>2</sub> and TiO<sub>2</sub>, n-type material), using the triple cation perovskite composition (CsFAMAPbIBr) and spiro-OMeTAD as HSC (p-type material). The photovoltaic parameters, obtained from the measurements of the current voltage (JV) curves are shown in Figure 1 and listed in Table 1. For clarity, we just show the results from day 1 (fresh) and day 28 (aged). Figure S1 summarises the photovoltaic parameters of the different measurement days. During the whole duration of the experiment, the devices were kept in dark and under dry air conditions (RH < 10% H<sub>2</sub>O), obtained via a synthetic airflow.

For fresh devices, the J-V hysteresis is more accentuated for devices with TiO<sub>2</sub> as ESC. This might be attributed to the higher electron collection efficiency and better band alignment of energy levels of SnO<sub>2</sub>, which helps to decrease and even eliminate hysteresis [19]. However, in both cases, a reduction in the JV hysteresis is observed after keeping the solar cells in dark and dry air conditions (Figure 1 and Table 1). In previous publications, we and others have reported this kind of behaviour [18,20,21]. It has been suggested that O<sub>2</sub> might have two major effects over this perovskite configuration. First, spiro-OMeTAD becomes oxidised, which helps to increase its conductivity, so getting a better charge extraction, and also to lower the HOMO value, which results in better band alignment, reducing carrier losses due to band mismatching [20]. The better charge extraction properties and reduced losses produce better values of FF and  $V_{\infty}$  in the aged solar cells. This process happens in the very early days after the storage of the devices, being clearly observable in the results on day 4 (Figure S1). Despite this increase in the very early days, these devices still present a high degree of hysteresis, and we have observed that the storage in dry air conditions over long periods of time helps to reduce it [17,18]. With longer aging times, device hysteresis reduces, obtaining higher values of  $V_{\infty}$  and FF. As well, the difference between the Voc values in the forward sweeps (from Jsc to Voc) and reverse sweeps (from Voc to Jsc) tends to be smaller ( $\Delta$ Voc = Voc (Rev) – Voc (fwd);  $\Delta$ Voc(fresh,  $TiO_2$  = 134 mV versus  $\Delta Voc(aged, TiO_2)$  = 23 mV). We observe significant changes in Voc, increasing 85 mV for TiO<sub>2</sub>-based devices and 70 mV for SnO<sub>2</sub>-based devices. This important improvement might be associated with an improved carrier recombination at the interfaces, which greatly affects Voc. Metal halide perovskite are both ionic and electronic semiconductors, and the presence of mobile ions at the interfaces were appointed as the main cause for hysteresis in perovskite solar cells [13]. Thus, the reduction of hysteresis in our devices is related to a reduction in the influence of mobile ions over recombination. It has been previously suggested that the O<sub>2</sub> molecules might also have a passivation effect on the perovskite lattice [13,21,22]. They might occupy some of the halide defects in the perovskite lattice, improving both ionic mobility and non-radiative carrier recombination. Using Voc stabilization, it is possible to track the influence of ionic mobility over carrier recombination, as this technique correlates a slow Voc stabilization with a slow ionic redistribution that leads to an increase interfacial recombination, and, thus, reduced Voc [23].

Another important feature is that SnO<sub>2</sub>-based devices do not show any change in the Jsc, while we observed a reduction in TiO<sub>2</sub>-based devices. This is related to TiO<sub>2</sub> causing photo-instability in the perovskite as it absorbs in the UV region [24,25], producing a slight degradation in the perovskite layer. Therefore, both from performance and from stability, we can confirm that SnO<sub>2</sub> works better than TiO<sub>2</sub> as ESC in n-i-p perovskite solar cells.

**Table 1.** Summary of the photovoltaic parameters for CsFAMAPbIBr on either TiO<sub>2</sub> or SnO<sub>2</sub> layers measured in fresh (1 day) and aged (28 days) conditions.

ESC/Perovskite	Aging Days	Sweep	Jsc [mA/cm <sup>2</sup> ]	Voc [V]	FF	PCE [%] (Champion Cell)
TiO2/CsFAMAPbIBr	1	Forw	$21.7 \pm 1.1$	$0.930 \pm 0.011$	$0.53 \pm 0.01$	10.7 ± 0.4 (11.5)
		Rev	$22.1 \pm 1.1$	$1.054 \pm 0.029$	$0.71 \pm 0.01$	16.6 ± 0.9 (18.4)
	28	Forw	$20.5 \pm 1.1$	$1.116\pm0.018$	$0.67 \pm 0.05$	15.3 ± 1.7 (16.8)
		Rev	$20.5 \pm 0.3$	$1.139 \pm 0.011$	$0.70 \pm 0.02$	16.3 ± 0.6 (17.0)
SnO2/CsFAMAPbIBr	1	Forw	$22.0\pm0.9$	$1.062 \pm 0.013$	$0.50 \pm 0.02$	11.8 ± 0.5 (12.5)
		Rev	$21.8\pm0.9$	$1.081 \pm 0.022$	$0.66 \pm 0.01$	15.6 ± 1.0 (17.9)
	28	Forw	$21.6\pm0.9$	$1.146 \pm 0.006$	$0.71 \pm 0.04$	17.5 ± 1.0 (19.6)
		Rev	$21.8\pm0.7$	$1.151 \pm 0.005$	$0.70 \pm 0.02$	17.5 ± 0.6 (18.6)

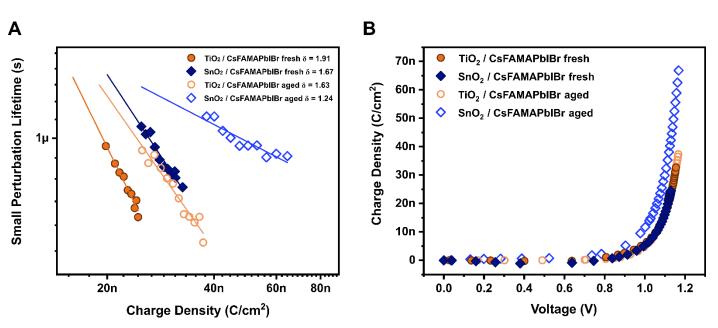


**Figure 1.** JV curves measured at 1 Sun conditions (100 mW/cm<sup>2</sup>, AM 1.5 G) in forward (dashed line) and reverse (solid line) scan for CsFAMAPbIBr on SnO<sub>2</sub> (**A**) and TiO<sub>2</sub> (**B**) in fresh and aged devices.

Then, we moved onto the study of the dependence of the ESC choice on the carrier recombination. For doing this, we employed a set of optoelectronic transient techniques, transient photovoltage (TPV) and transient photocurrent (TPC), which we have widely used in previous studies on perovskite solar cells, as well as on organic solar cells or dye sensitised solar cells [11,15,30]. Both techniques are small-perturbation techniques based on the application of a laser pulse that generates a small perturbation in voltage while using different background illumination intensities, which allow registering the experiment at different voltages. With TPV, we study the time that it takes for the voltage to return to its steady state, which depends on the background bias, and it is correlated with the recombination in the solar cell. TPC allows us to determine the number of charges that the laser pulse generates in the device. By combining both techniques, we can determine the charge density in the device with a technique termed as differential capacitance (DiffCap).

We study the recombination scenario with the small perturbation lifetimes obtained from the TPV decays at stable photovoltages. As it has been previously highlighted, at low-light intensities (low values of voltages), the lifetimes obtained from TPV are affected by the discharge of the ionic layer accumulated at the perovskite interfaces, which is the ionic redistribution once the device is illuminated [10,17]. Therefore, we should compare the lifetimes at high voltage values (Figure S2), which we can identify as the region in which carriers accumulate in the bulk of the perovskite (Figure S3). For this reason, we compare the recombination kinetics at different carrier densities (Figure 2A), obtained from the differential capacitance method (Figures 2B and S3). In this way, we compare the kinetics when the contacts are depleted with charges, and they start to accumulate exponentially in the bulk [15]. This part is easily distinguishable in the analysis of the charge density as a function of voltage (Figure S3), where two different regimes are distinguished. First, a linear increase with voltage of the charge at lower values of voltage. Then, after the contacts are depleted with charges, they accumulate in the bulk of the perovskite, which we can observe as an exponential increase with voltage [15]. In these samples, as expected, we observe slower lifetimes for the aged devices (Figure 2A), which confirm that the increase in Voc is associated with a decrease in the recombination scheme of the system. This is also supported by the recombination orders in the perovskite. The small perturbation lifetime follows a power law dependence with the charge density  $\tau_{\Delta n}$  =

 $\tau_{\Delta n_0}(Q/Q_0)^{-\lambda}$ , which is used to calculate the recombination order ( $\delta$ ), ( $\delta = \lambda + 1$ ) [26]. All the devices are ruled by first and second order ( $\delta = 1-2$ ) corresponding to the trapassisted and surface recombination and band-to-band free carriers [27]. However, a clear decrease in the recombination order of the samples is observed after the aging treatment. Again, we associated it with a reduced carrier recombination at the interfaces because of the reduced ionic activity and motion. Using the recombination orders, it is possible to estimate the total carrier lifetime, or a pseudo-first-order lifetime of total excess charge carries via  $\tau = \tau \Delta n \delta$  (Figure S4). We observe very similar trends independently of the ESC employed.



**Figure 2.** (**A**) Carrier lifetime obtained from TPV decays as a function of the carrier density after the subtraction of the geometric capacitance. (**B**) Carrier density obtained from the integration of the capacitance and after the subtraction of the geometric capacitance. The orange circles represent the TiO<sub>2</sub>/CsFAMAPbIBr, and blue diamonds represent the SnO<sub>2</sub>/CsFAMAPbIBr. In both cases, the filled symbols represent the fresh samples and the empty symbols – the samples after the aging treatment.

Next, we study the ionic influence over carrier recombination. For this purpose, we employed a variation of TPV, known as transient of the transient (TROTR), previously used by our research group and others [14,16,18]. In this technique, TPV transients are registered at the early stages of illumination, prior to Voc stabilization, where the ionic population is not fully distributed in the perovskite layer [17,18]. Figure 3 shows the Voc stability over time for fresh and aged devices. Here, we clearly observe a slower stabilization of voltage in illumination conditions for fresh devices mainly due to its higher hysteresis, which agrees well with results of previous studies [18].

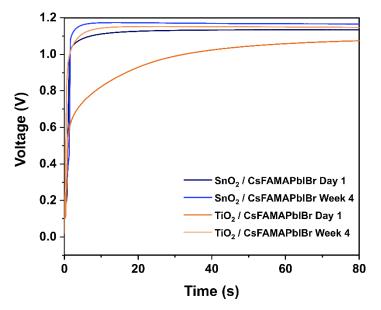
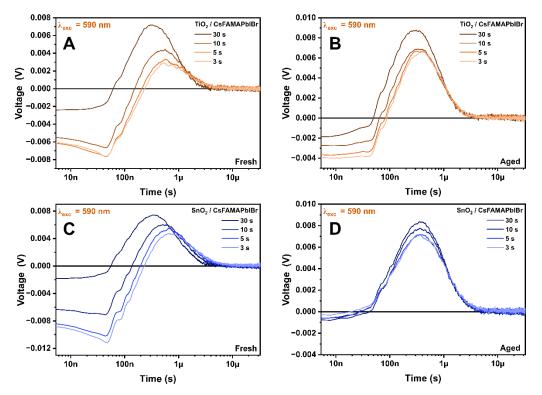


Figure 3. Voc stability over time for CsFAMAPbIBr on TiO<sub>2</sub> and SnO<sub>2</sub> for fresh and aged devices.

This slow V $\infty$  stabilization is associated with the ionic redistribution in the perovskite layer, a slow process that affects carrier recombination, inducing slower V $\infty$  values until ions redistribute in the bulk of the perovskite [13,16]. During this time, when TPV transient pulses are registered – at 3, 5, 10 and 30 s – a negative spike appears. This negative transient, associated with an increased recombination at the interfaces is observable in every fresh device (see Figure 4A,C). Then, after the aging treatment (see Figure 4B,D), this spike is reduced for the case of TiO<sub>2</sub>, or it even disappears, as with SnO<sub>2</sub>-based devices and, as previously observed, after aging for 45 days [17]. These results suggest that the ionic accumulation at the interfaces enhances recombination, most likely, trap-assisted carrier recombination, thus yielding lower values of Voc. After storing the devices under dry air conditions, the ionic influence over carrier recombination is reduced, which suggests that O<sub>2</sub> might have a passivation effect over carrier traps. However, since the hysteresis is not fully negligible for aged devices, they still present the negative transient signal attributed to the ionic movement. Moreover, there are differences between TiO<sub>2</sub> and SnO<sub>2</sub> aged that can be due to the better properties for SnO<sub>2</sub> as ESC and the differences between mesoporous and planar structure, showing that planar structure avoids better the accumulation of carriers at the ESC/perovskite interface [28].



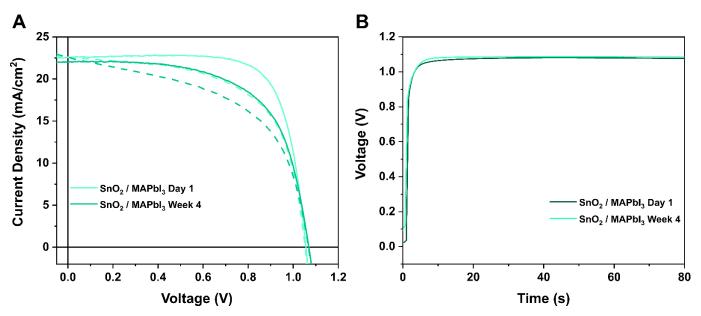
**Figure 4.** TPV transients in logarithm scale obtained at different illumination times at 1 Sun intensity for fresh and aged CsFAMAPbIBr on TiO<sub>2</sub> (**A**,**B**) and SnO<sub>2</sub> (**C**,**D**) ESC.

#### 3.2. CsFAMAPIBr<sub>3</sub> vs. MAPbI<sub>3</sub>-Based Devices on SnO<sub>2</sub> Layer

Finally, as we have already observed in previous studies, perovskite composition also plays an important role in the dynamics of ionic and charge recombination [18,21]. Therefore, we carried out the same studies on MAPbI<sub>3</sub> fabricated on SnO<sub>2</sub> and compared them with those of the triple cation perovskite shown in Section 3.1. JV curves show that hysteresis after aging SnO<sub>2</sub>/MAPbI<sub>3</sub> for 28 days was not reduced as it occurs for SnO<sub>2</sub>/CsFAMAPbIBr devices, showing a less positive effect from the aging process (see Figures 5A and S5, Table 2). There is an improvement in the fill factor occurring in the first days, but not in the V<sub>oc</sub> values, contrary to what we have observed for the triple cation perovskite composition. This reflects that carrier extraction might improve, probably due to the spiro-OMeTAD oxidation, with improvements in both conductivity and band alignment [17,29], but carrier recombination might increase, as we observe even smaller values of V<sub>oc</sub> over the period of the experiment. The effect of aging on MAPbI<sub>3</sub> was also confirmed in the V<sub>oc</sub> stability over time shown in Figure 5B, where, for both fresh and aged devices, the time to reach the maximum V<sub>oc</sub> was almost the same (approximately 10 s).

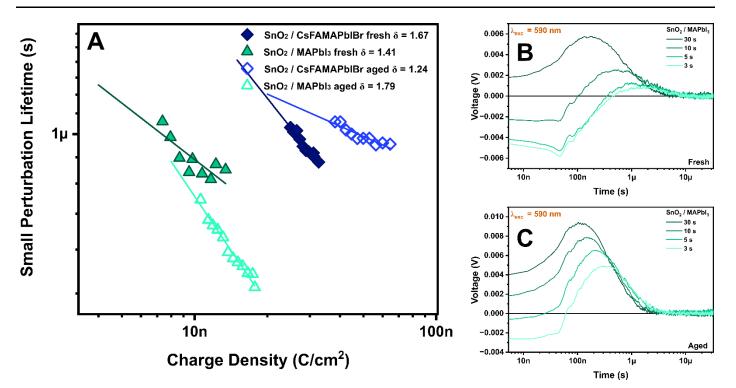
These observations on the photovoltaic parameters might suggest that not all the perovskite compositions are valid for this dry air aging treatment. In previous publications, we already observed that the use of bromine-based compositions instead of iodine were more prone to improve the photovoltaic parameters [30]. These results make us think that oxygen cannot passivate any defects at the MAPbI<sub>3</sub> perovskite lattice, as we do not observe any long-term improvement in the photovoltaic performance.

Then, we evaluate the carrier recombination kinetics for this composition using TPV and TPC. One thing that is clear is that with MAPbI<sub>3</sub>, there is a greater influence of the geometric capacitance on the differential capacitance technique, as it is shown in Figure S6, which might be attributed to an increased ionic accumulation at the interfaces [15,31]. This bigger influence can lead to a wrong interpretation of the carrier recombination scheme (Figure S7). Therefore, to perform a correct analysis of the different perovskite compositions, we remove the geometrical capacitance component (Figures 6A and S7). After doing this, MAPbI<sub>3</sub> presents a faster carrier recombination than CsFAMAPbIBr, which would explain the lower values of  $V_{\infty}$  after the aging treatment. Although the recombination orders for SnO<sub>2</sub>/MAPbI<sub>3</sub> were in the same order as those for CsFAMAPbIBr devices ( $\delta = 1-2$ ), the recombination increased after 28 days instead of decreasing as for the other devices. Despite having discarded any benefits from oxygen exposure for MAPbl<sub>3</sub> from the JV and Voc stability, we also measured TROTR (Figure 6B,C). Here, after 28 days of oxygen exposure, we observe a similar behaviour to TiO2-based CsFAMAPbIBr devices, as the negative component remains, but it is not as intense as in the fresh samples. While we ruled out the possibility of this being related to any changes in the bulk of the perovskite, it could be associated with the use of  $SnO_2$  as ESC. The higher conductivity and smaller surface area of SnO<sub>2</sub> compared to the use of mesoporous TiO<sub>2</sub> can have a positive effect on how ions move from this interface. However, as MAPbI<sub>3</sub> JV curves did not improve, we continue observing the negative spike at very early timescales after the laser pulse. Ionic accumulation at the interfaces was described as the major influence for the great capacitance values reported for perovskites [9,31]. Moreover, recently, it was described that it is iodine defects the main responsible for such ionic accumulation. A dditionally, these defects present a lower formation energy and higher defect density for MAPbl<sub>3</sub> compared to the triple cation composition [32]. Therefore, the higher ionic density at the interfaces of MAPbI3 devices leads to higher capacitance values, increased carrier recombination, and finally, a higher degree of hysteresis in the devices. All the ionic components might contribute to the ionic capacitance observed. As well, it has been previously suggested that MA<sup>+</sup> is the main responsible for the hysteresis observed in the JV curves. However, its ionic mobility is very slow compared to iodide and its vacancies, which may suggest that we cannot see any effect on the TROTR, but it is still observable in JV hysteresis [33,34].



**Figure 5.** JV curves measured at 1 Sun conditions (100 mW/cm<sup>2</sup>, AM 1.5G) in forward (dashed line) and reverse (solid line) scan for fresh and aged samples of MAPbI<sub>3</sub> deposited on SnO<sub>2</sub> (**A**) and Voc stability over a period of time for fresh and aged MAPbI<sub>3</sub> samples (**B**).

ESC/Perovskite	Aging Days	Sweep	Jsc [mA/cm²]	<b>V</b> oc <b>[V]</b>	FF	PCE [%] (Champion Cell)
SnO2/MAPbI3	Day 1	Fwd	$21.2 \pm 0.8$	$1.072 \pm 0.007$	$0.56 \pm 0.02$	12.7 ± 0.6 (13.6)
		Rev	$20.5 \pm 0.9$	$1.076 \pm 0.006$	$0.64 \pm 0.01$	14.2 ± 0.6 (15.0)
	Week 4	Fwd	$21.2 \pm 0.9$	$1.058 \pm 0.013$	$0.64 \pm 0.02$	14.2 ± 0.8 (15.5)
		Rev	21.0 ± 0.9	$1.065 \pm 0.007$	$0.71 \pm 0.02$	16.0 ± 1.1 (17.7)



**Figure 6.** Carrier lifetime obtained from TPV decays as a function of the carrier density after the subtraction of the geometric capacitance for SnO<sub>2</sub>/CsFAMAPbIBr and SnO<sub>2</sub>/MAPbI<sub>3</sub> samples (**A**). TPV decays obtained during different times of illumination for fresh (**B**) and aged (**C**) SnO<sub>2</sub>/MAPbI<sub>3</sub>.

#### 4. Conclusions

In conclusion, CsFAMAPbIBr perovskite-based devices showed different performance when using either SnO<sub>2</sub> or TiO<sub>2</sub> as ESC. SnO<sub>2</sub> as ESC showed higher electron collection efficiency and greater stability than TiO<sub>2</sub>. However, when comparing fresh and aged devices, both ESC reveal a decrease in recombination, being the aging process favourable for the V<sub>oc</sub> enhancement for all the CsFAMAPbIBr devices. These results suggest that, after storing the devices under dry conditions, O<sub>2</sub> presents two main effects: the better extraction of charges because of the higher oxidation of spiro-OMeTAD, and the decrease in recombination due to its passivation effect over carrier traps. Interestingly, for the MAPbI<sub>3</sub> perovskite devices, the aging process did not have the same results. The hysteresis was almost identical for fresh and aged devices and with higher recombination after the exposition of the devices. This clearly shows that perovskite composition plays

**Table 2.** Summary of the photovoltaic parameters for MAPbI<sub>3</sub> on SnO<sub>2</sub> measured in fresh (1 day) and aged (28 days) conditions.

an important role in the dynamics of ionic and charge recombination, with MAPbI<sub>3</sub> having a greater ionic accumulation compared to CsFAMAPbIBr that leads to faster carrier recombination, and, consequently, worse device performances.

Supplementary Materials: The following supporting information can be downloaded at: www.mdpi.com/xxx/s1, Figure S1: Photovoltaic parameters of an average of 10 devices measured on fresh devices, and after 4 days, 1, 2, 4 (28 days), and 6 weeks for (A) SnO<sub>2</sub>/CsFAMAPbIBr, (B) TiO<sub>2</sub>/CsFAMAPbIBr; Figure S2: Small perturbation lifetimes obtained from the exponential fitting of the TPV decays as a function of the light bias for the CsFAMAPbIBr samples deposited on top of TiO<sub>2</sub> (orange circles) or SnO<sub>2</sub> (blue diamonds) as ESC. The samples were measured fresh (filled symbols) and after the aging treatment (empty symbols); Figure S3: Capacitance obtained from the DiffCap method at different voltages (A) and after subtracting the geometric capacitance (charges accumulated at the contacts) (B). Charge density obtained at different voltages after the integration of the capacitance obtained with the DiffCap method, including carriers in the bulk and in the contacts (C). Charge density accumulated in the perovskite bulk (D). Small perturbation lifetime as a function of the total charge density accumulated in the device (charges in the bulk and in the contacts) (E). Small perturbation lifetime as a function of the charge density accumulated in the bulk of the perovskite. The lines represent the power law fitting  $\tau_{\Delta n} = \tau_{\Delta n_0} (Q_{Q_0})^{-\lambda}$  used to calculate the recombination order ( $\lambda = 1, j \neq 1$ ). (if the last is  $T_{\Delta n} = \tau_{\Delta n_0} (Q_{Q_0})^{-\lambda}$ recombination order ( $\delta = \lambda + 1$ ) of the devices (F); Figure S4: Total carrier lifetime ( $\tau = \tau_{\Delta n} \cdot \delta$ ) as a function of the charge density stored in the bulk of the perovskite; Figure S5: Photovoltaic parameters of an average of 10 devices measured on fresh devices, and after 4 days, 1, 2, 4 (28 days), and 6 weeks for SnO<sub>2</sub>/MAPbI<sub>3</sub>; Figure S6: Capacitance obtained from the DiffCap method at different voltages (A) and after subtracting the geometric capacitance (charges accumulated at the contacts) (B).

Charge density obtained at different voltages after the integration of the capacitance obtained with the DiffCap method, including carriers in the bulk and in the contacts (C). Charge density accumulated in the perovskite bulk (D); Figure S7: Small perturbation lifetimes obtained from the exponential fitting of the TPV decays as a function of the light bias (A) for the CsFAMAPbIBr (blue) and MAPbI<sub>3</sub> (green) samples desposited on top of SnO<sub>2</sub> as ESC. The samples were measured fresh (filled symbols) and after the aging treatment (empty symbols). Small perturbation lifetime as a function of the total charge density accumulated in the device (charges in the bulk and in the contacts) (B). Small perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function of the charge density accumulated in the perturbation lifetime as a function

Small perturbation lifetime as a function of the charge density accumulated in the bulk of the perovskite. The lines represent the power law fitting  $\tau_{\Delta n} = \tau_{\Delta n_0} (Q/Q_0)^{-\lambda}$  used to calculate the recombination order ( $\delta = \lambda + 1$ ) of the devices (C). Total carrier lifetime ( $\tau = \tau_{\Delta n} \cdot \delta$ ) as a function of the charge density stored in the bulk of the perovskite (D).

**Author Contributions:** Conceptualization, J.J.-L. and E.P.; methodology J.J.-L. and E.P.; validation J.J.-L., M.M. and E.P.; writing—original draft, review and editing J.J.-L., M.M. and E.P.; supervision E.P. All authors have read and agreed to the published version of the manuscript.

Funding: This research was funded by MINECO (project PID2019-109389RB-I00) and SGRA-GAUR2017SGR00978.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Acknowledgments: The authors acknowledge funding from MINECO and the financial support from ICIQ, CERCA, and ICREA.

Conflicts of Interest: The authors declare no conflict of interest.

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